

WJEC Wales Physics GCSE

RP10: Radioactive Decay Practical Notes

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Practical 10: Determination of the half-life do a model radioactive source

Equipment

- 50x cubes with one face shaded
- Plastic tub
- Tray

Diagram



Image: WJEC

Method

- 1. Put the cubes into the tub and shake it gently.
- 2. Throw the cubes from the tub into the tray and record the number that land shaded face up.
- 3. Remove these cubes from the tray.
 - These are considered to be decayed atoms.
- 4. Put the remaining cubes back in the tub, shake, and throw onto the tray again, recording the number that land shaded face up and then removing them from the group.
- 5. Repeat this process 10 times in total.
- 6. Plot a graph of the number of cubes/atoms remaining against the number of throws.
 - Use this to determine the half-life of the cubes.
 - Remember that half-life is the time taken for the number of radioactive nuclei to halve.

▶ Image: Contraction PMTEducation

Tips

 Radioactive decay is a random process. The cubes that land shaded side up represent decayed atoms, whereas the remaining cubes represent those atoms that have not yet decayed.